

Analyse des couplages entre mélange, transfert et réaction dans les bioréacteurs

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Equipe : Transferts Interfaces Mélange

Journée : Utilisation de la CFD en génie des procédés.
Gestion de la complexité des systèmes

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Industrial Bioreactors

Large scale system : aerated, stirred, biological transformation

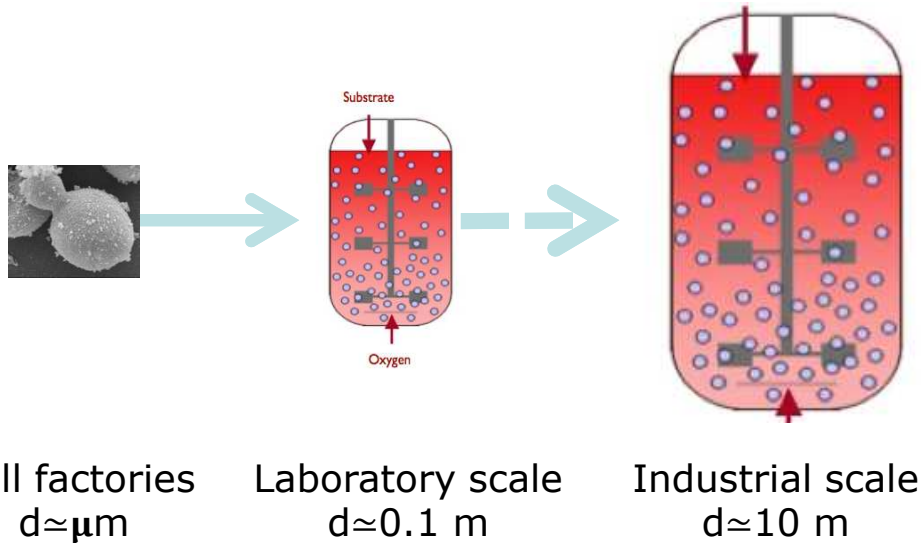


Industrial production of
Pharmaceuticals
Energy (Bio fuels)
Water





Modeling of Bioreactors



Industrial challenges:

- Design
- Performances

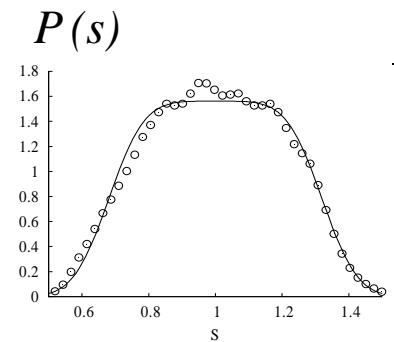
Academic challenges:

- Hydrodynamics
- Multiphase
- Mixing
- Mass transfer
- Heat transfer
- Bioreaction
- *Adaptation of microorganisms*

Strongly coupled phenomena
Large range of scales
Macro ^[1-5] and micro^[6-8] mixing issues

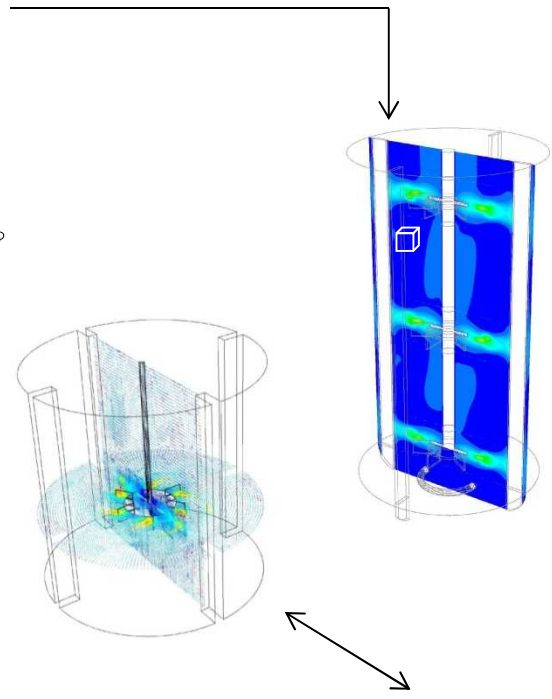
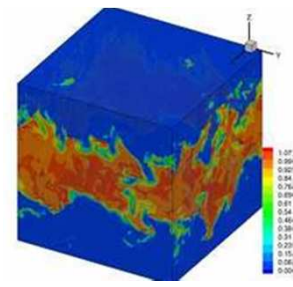
[1] Larsson et al., 1996; [2] Bylund et al., 1998; [3] Enfors et al., 2001 [4] Senn et al., 1994; [5] Lin and Neubauer, 2000
[6] Dunlop and Ye, 1990; [7] Al-Homoud and Hondzo, 2008; [8] Garcia et al., 2009

CFD Tools



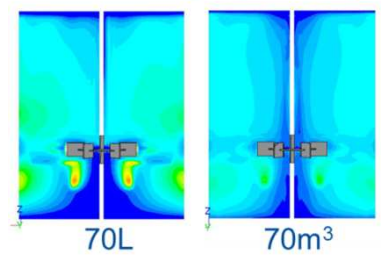
Concentration distribution

DNS Numerical experiments

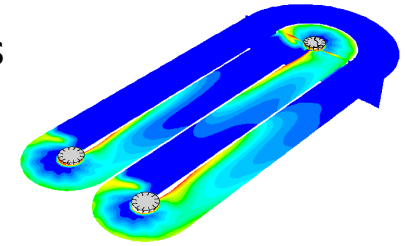


Industrial cases

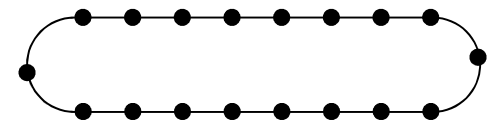
Rétention gaz (0-20 %)



Eulerian Multiphase Scale-up



1D Model $C(z,t)$



$$\frac{\partial C}{\partial t} + U \frac{\partial C}{\partial Z} - \frac{1}{Pe} \frac{\partial^2 C}{\partial Z^2} = R(C)$$

LES, RANS
Experimental validation
PIV
PIV-HF (IMFT, Fermat)

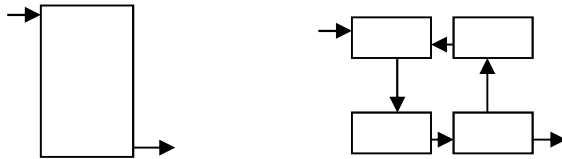


Limitations of existing approaches

Chemical Reactor Engineering

Ideal reactor ^[9] (CSTR, PFR)

Compartment Model ^[10]



Biological reaction model

Kinetic (unstructured, structured)

Metabolic

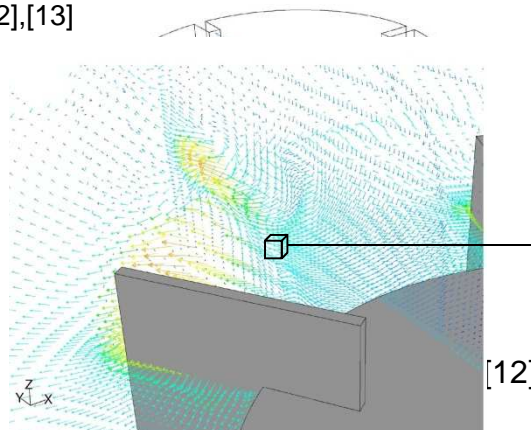
Uptake rate $\Phi(S)$ $g_S g_X^{-1} h^{-1}$

Computational Fluid Mechanics

RANS ^{[11],[12],[13]}

LES ^[12]

DNS ^{[14], [15]}

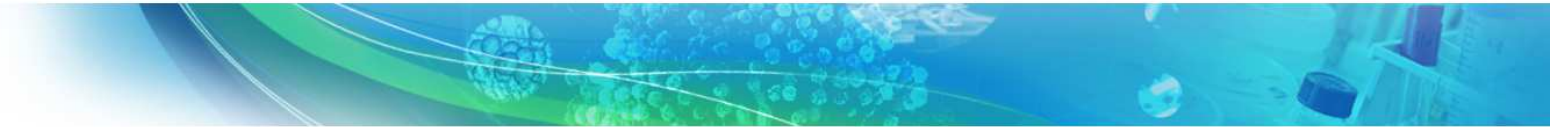


Coupling

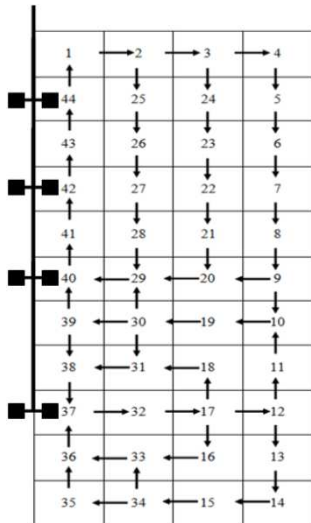
Conservation equation for the substrate over a *volume of control*

$$\frac{\partial S}{\partial t} + U \nabla S - \nabla (D \nabla S) = -\Phi(S) X$$

[9] Villiermaux, 1995; [10] Delvigne et al., 2005; [11] Schmalzriedt et al., 2003; [12] Delafosse et al., 2008; [13] Lapin et al., 2006. ; [14] Douaire 2010; [15] Linkès 2012



Should work but ...



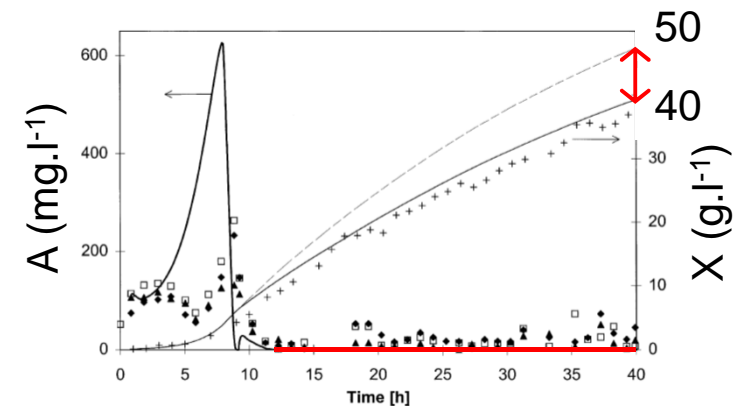
Compartment Model Approach (Vr̀abel et al. 2001) + Metabolic model *E. coli* (Xu et al. 1999)

Based on a known flow map $Q_{i,j}$

Cell concentration over predicted
By product formation underpredicted

Scale dependent Yield constants

$$Y_{\text{large scale}} = Y_{\text{lab scale}} * 0.75$$



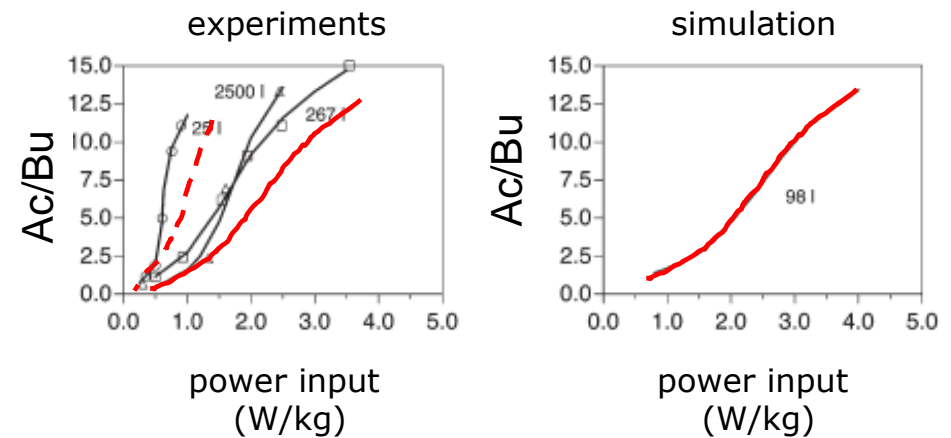
3D - Two phases CFD RANS (Schmalzdriedt et al. 2003)
+ Kinetic Model *B. subtilis* (Moes et al. 1985, Griot 1987)

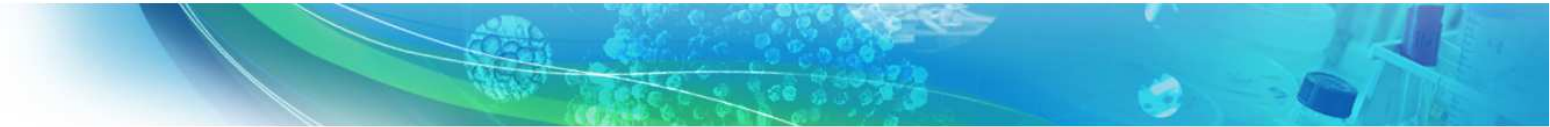
Increasing the power input favours Acetoïn production

→ Sensitivity to mixing/transfer

The simulated ratio is out of the expected range

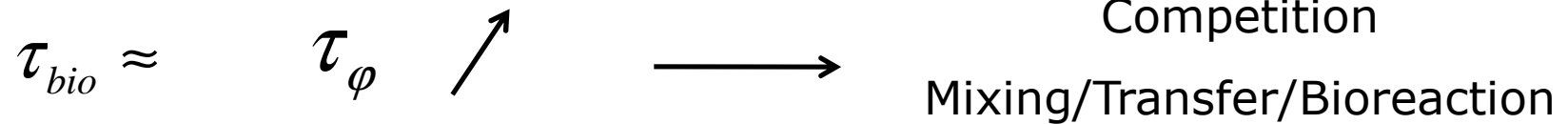
→ Kinetic model is scale dependent ...



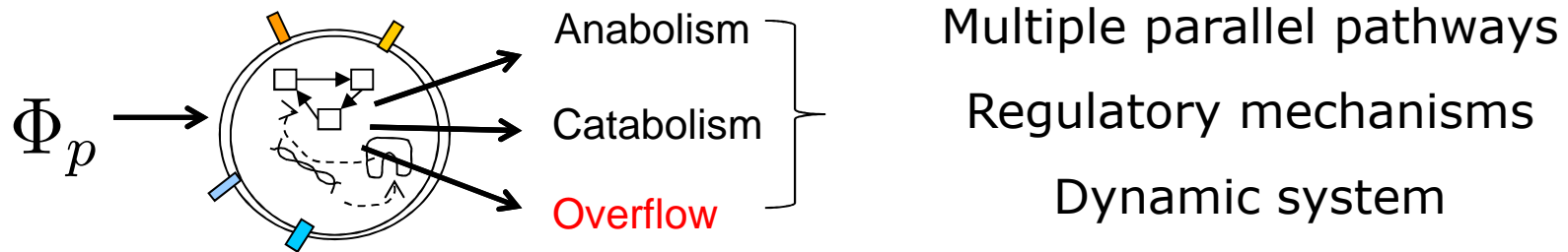


Restart Analysis

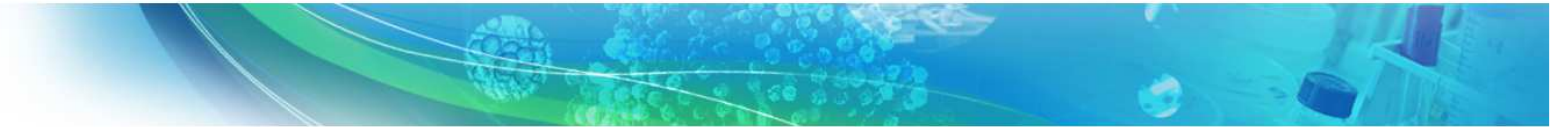
Through scale-up



Time scale estimation ?



Adaptation to concentration fluctuations experienced



Characteristic time scales of mixing

Table 1: Time (in seconds) and length scales of mixing in two bioreactors (lab-scale 3L, industrial scale 20 m³) equipped with a Rushton turbine. Average power input $\epsilon_V = 1.3 \text{ kW.m}^{-3}$ in both cases. t_C : macromixing time, t_S mesomixing time, t_E : micromixing time due to engulfment, t_{Ds} micromixing time due to diffusion, Λ : Taylor Macro scale, η_K : Kolmogorov scale, η_B : Batchelor scale

Mecanisms	Time scale	Length scale	Labscale	Industrial (1/0.05)
Macromixing	$t_C = \frac{V}{1.8 N_{QP} N D^3}$	$V^{1/3}$	0.8	13
Mesomixing	$t_S = 2 \left(\frac{\Lambda^2}{\epsilon} \right)^{\frac{1}{3}}$	$\Lambda = w \quad \Lambda = \frac{1}{2} \frac{k^{3/2}}{\epsilon}$	0.03	0.2/0.55
Micromixing (Engulfment)	$t_E = 17 \left(\frac{\nu}{\epsilon} \right)^{\frac{1}{2}}$	$\eta_K = \left(\frac{\nu^3}{\epsilon} \right)^{\frac{1}{4}}$	0.015	0.015/0.07
Micromixing (Diffusion)	$t_{Ds} = 2 \left(\frac{\nu}{\epsilon} \right)^{\frac{1}{2}} \text{arcsinh}(0.05 Sc)$	$\eta_B = \left(\frac{\nu D_m^2}{\epsilon} \right)^{\frac{1}{4}}$	0.008	0.008/0.0026

Macro and meso mixing times increase with reactor size

- A. Delafosse, PhD, 2008
- B. M. Linkès, M. Martins Afonso, P. Fede, J. Morchain, P. Schmitz, Numerical study of substrate assimilation by a microorganism exposed to fluctuating concentration, Chemical Engineering Science. 81 (2012) 8–19.



Characteristic time scales of bioreactions

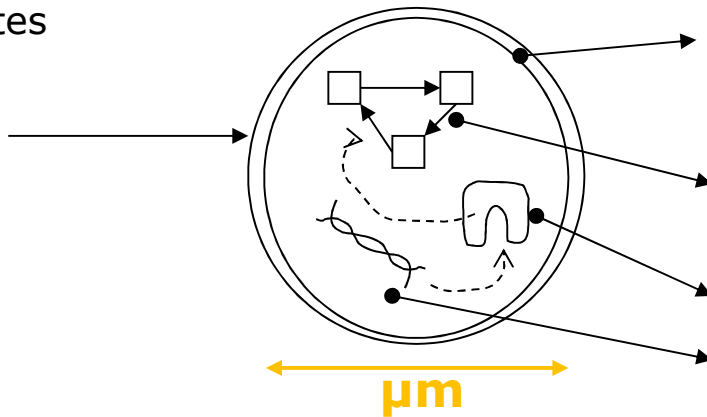
$$\frac{dC}{dt} = \frac{C}{\tau_C}$$

Substrates

C

N

O₂



CO₂

Products : Excreted Metabolites (1 à 100 s)

Intracellular Metabolites(→X, 1000 s)

Enzymes

Genes

Cell

$$\frac{dX}{dt} = \mu_{max} \frac{S}{K_S + S} X,$$

$$\tau_X = \frac{K_S + S}{\mu_{max} S},$$

Growth: 1000 s

Substrate

$$\frac{dS}{dt} = -q_{max} \frac{S}{K_S + S} X.$$

$$\tau_S = \frac{K_S + S}{\mu_{max} X} \approx \frac{S}{\mu_{max} X}$$

Uptake : 1-10 s

Growth and Uptake obey to different time scales
Uptake characteristic time decrease as the cell concentration increases



Mixing times and substrate uptake time

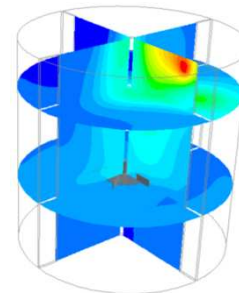
$$\tau_s = \frac{K_s + S}{\Phi_{\max} X} \quad X(t) \nearrow$$

Macromixing

$$\tau_M / \tau_s$$

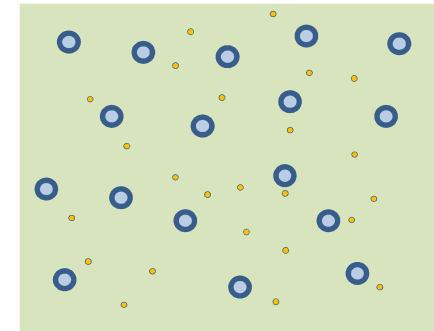
Homogeneous Reactor (Perfectly mixed)

Macroscopic gradients



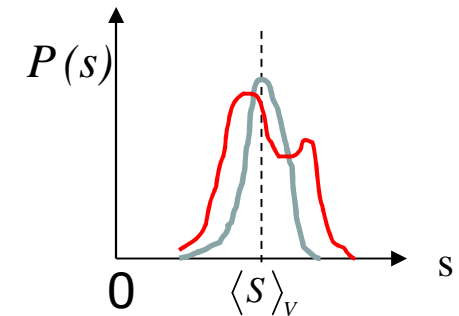
Mesomixing

Local concentration distribution



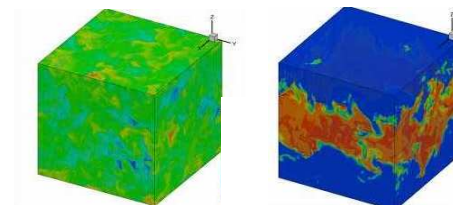
$$C_k \equiv \{c_k(\mathbf{x}, t)\}_k$$

$$\varepsilon_k \equiv \langle \alpha_k(\mathbf{x}, t) \rangle_V$$



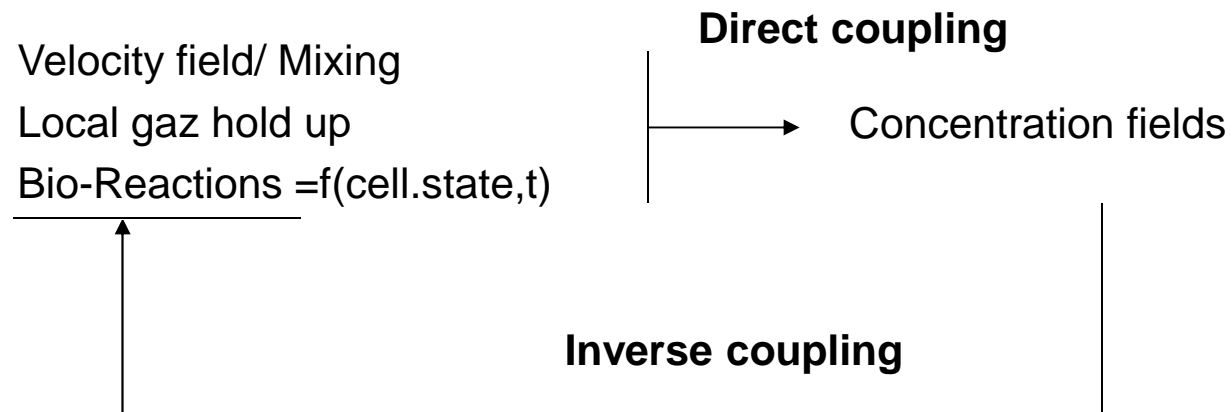
$$\tau_m \ll \tau_s$$

Non premixed reactants near the feed point





Specificity of biological systems : Two-way coupling

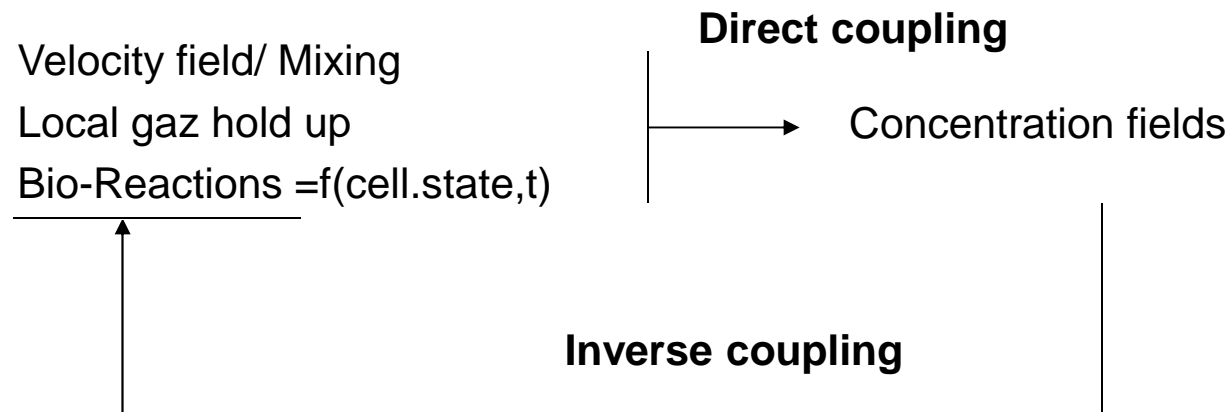


Is the cell at equilibrium with the liquid phase ?

Is the « average cell » approach sufficient ?

Is the concentration homogeneous within the volume of control ?

Specificity of biological systems : Two-way coupling



3D hydrodynamics and transfer : Multiphase CFD

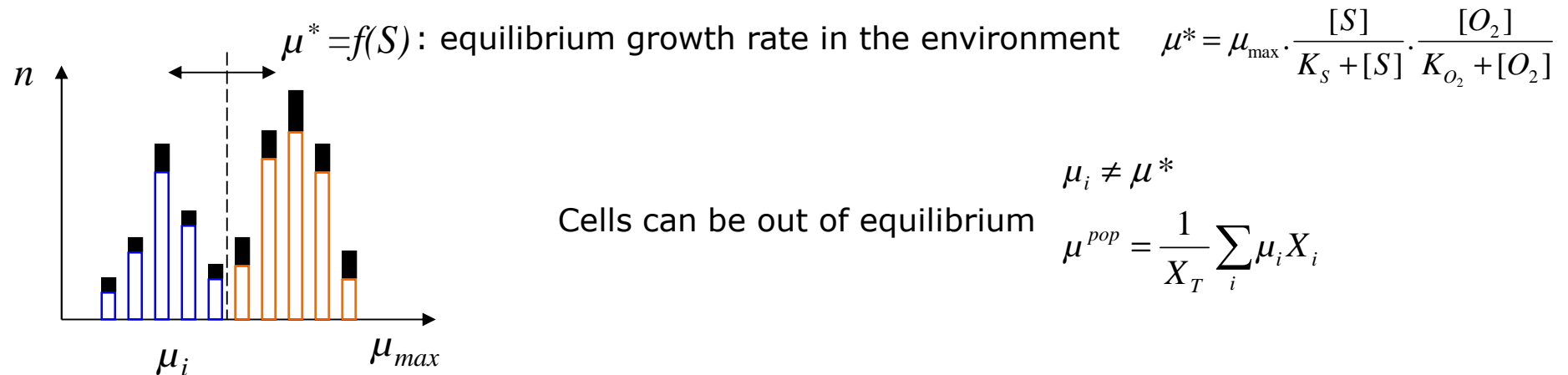
Equilibrium : Local reaction rates =f(local concentrations, *cell state*)

Describe and couple physical and biological heterogeneities :

PBM for Biological populations in an Eulerian framework



$$\frac{\partial n(\xi, x, t)}{\partial t} + u_i \frac{\partial n(\xi, x, t)}{\partial x_i} - \frac{\partial}{\partial x_i} \left(\Gamma \frac{\partial n(\xi, x, t)}{\partial x_i} \right) = - \frac{\partial}{\partial \xi} (u_\xi n(\xi, t)) + h(\xi, t)$$



Rate of change in the number of individuals in each class

- growth (within each class) $h_i = \mu_i$

- adaptation

$$u_\xi^A = \left(\frac{1}{T_A} + \mu_i \right) \frac{\mu^* - \mu_i}{\mu_{\max}}$$

- limitation

$$u_\xi^L = \left(\frac{1}{T_A} + \mu_i \right) \frac{\mu_i - \mu^*}{\mu_{\max}}$$

Population heterogeneity

Inertia

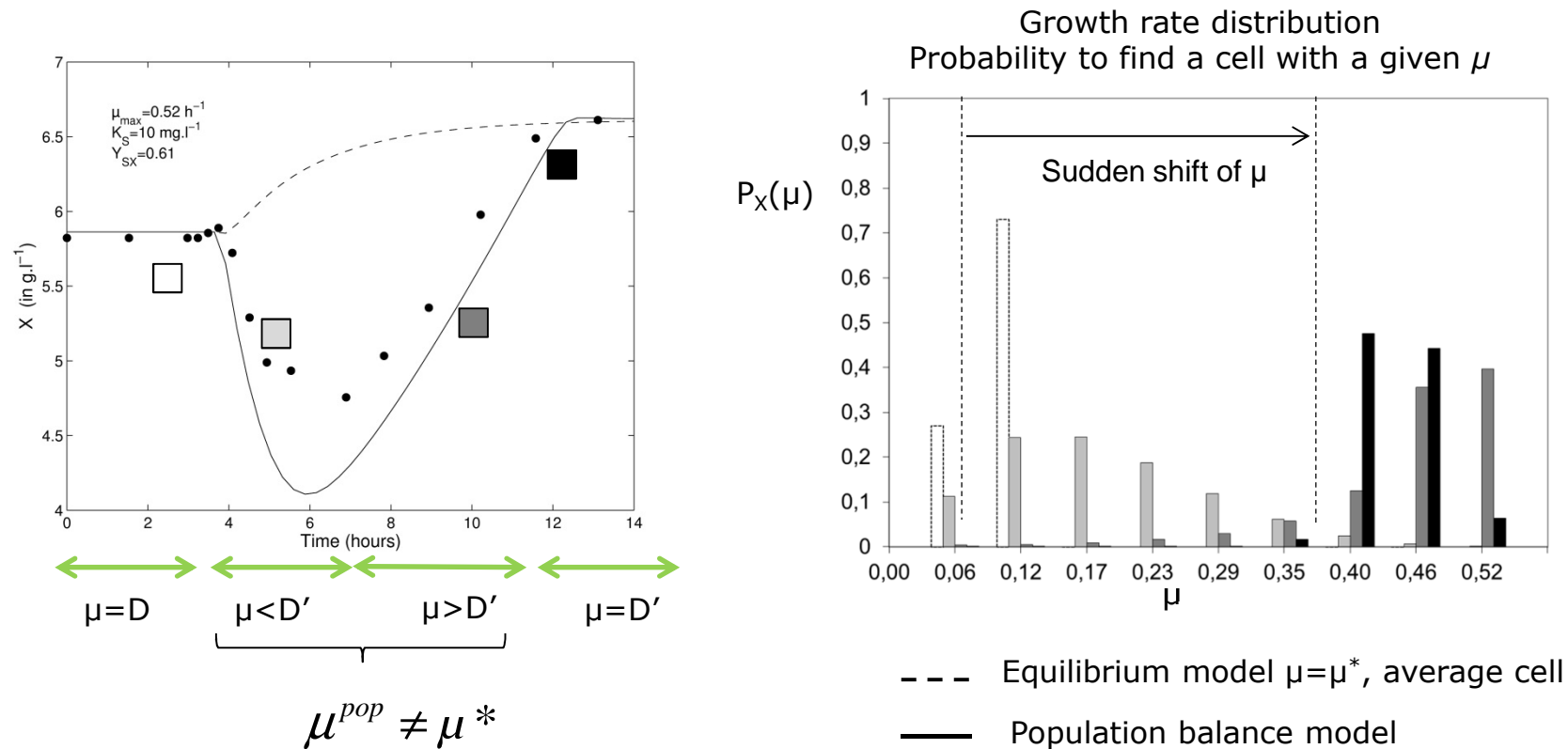
ξ : mass, size, intracellular component, reaction rate

$$T_A \approx \frac{1}{\mu_{\max}} \approx 1h$$



PBM → Out of Equilibrium ...

Experimental evidence: shift of the dilution rate ($D=Q/V$), perfectly homogeneous reactor
 $D=0.1 \text{ h}^{-1}$ to $D'=0.4 \text{ h}^{-1}$ at $t = 4 \text{ h}$ (Kätterer et al. 1986)

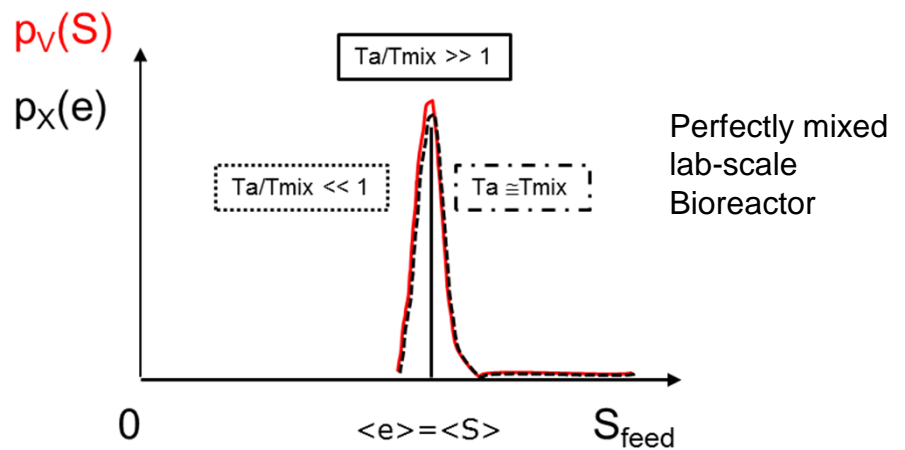
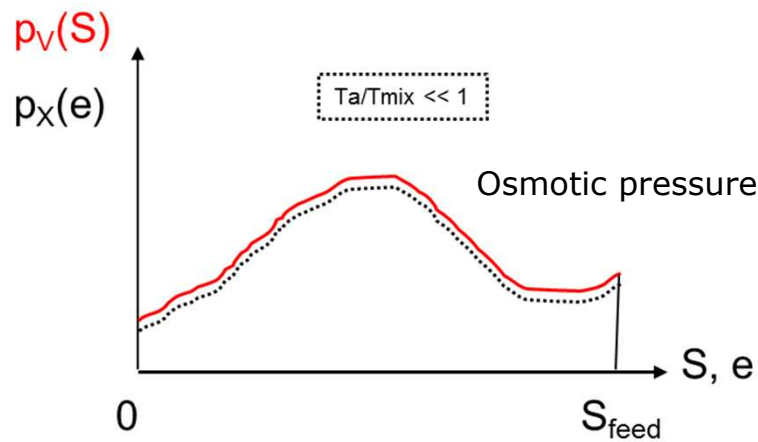
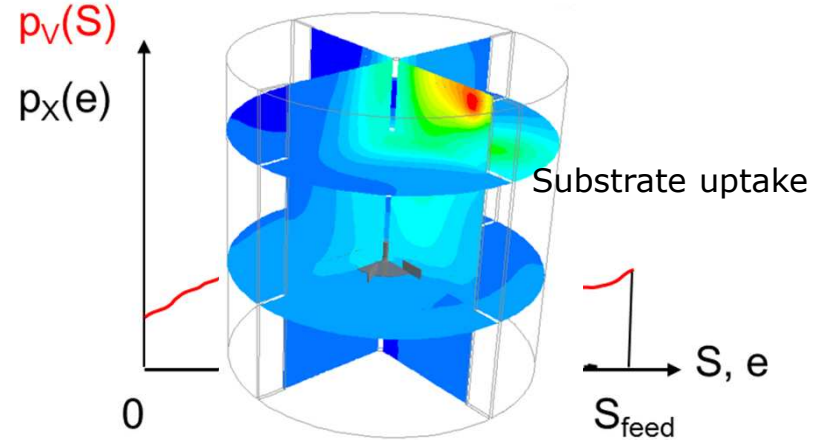
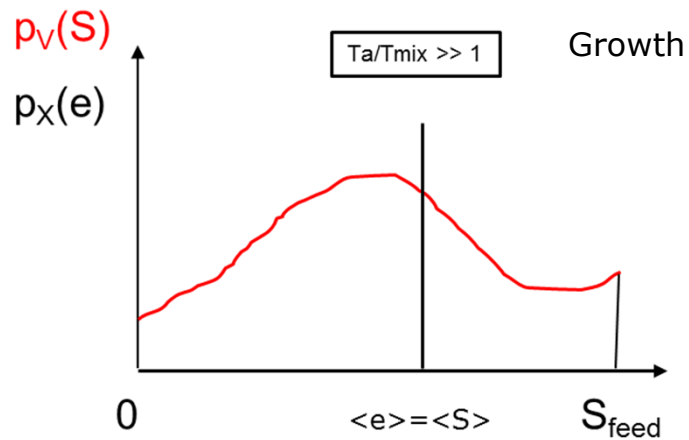


Morchain J, Fonade C. AIChE Journal. 2009, Morchain et al. AIChE Journal 2012

$P_V(S)$: probability to find a volume of liquid at the concentration S

$P_X(e)$: probability to find a cell at the internal concentration e

Time scale ratio and consequences



Effect of feed point location on population disequilibrium

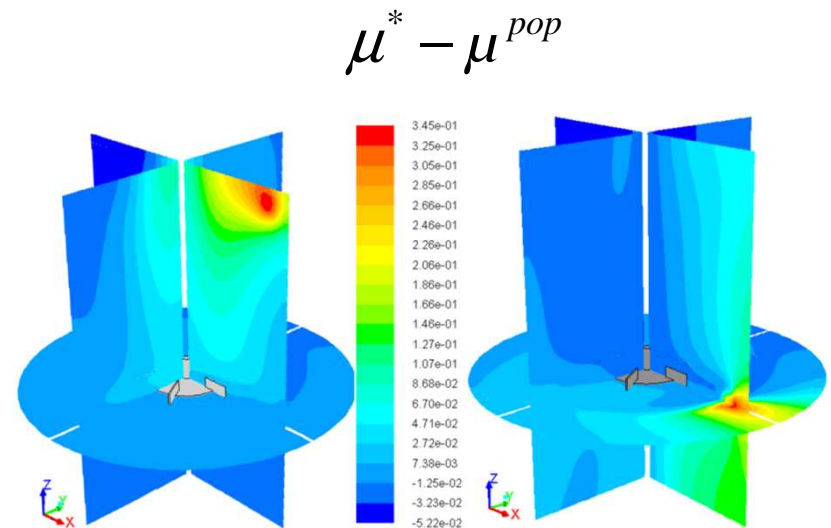
70 m³ aerated fermentor, kinetic reaction, G-L transfer

$\tau_M / \tau_S \rightarrow$ macroscopic gradients

1°) Growth ?

$$\mu^* = \mu_{\max} \cdot \frac{[S]}{K_S + [S]} \cdot \frac{[O_2]}{K_{O_2} + [O_2]} \quad \text{Spatially heterogeneous}$$

$$Ta = T_{\text{growth}} \gg T_{\text{mix}} \rightarrow \mu^{\text{pop}} \quad \text{Spatially homogeneous}$$



Local differences between the equilibrium growth rate based on local concentrations and the actual population growth rate

but....

$$\langle \mu^* \rangle = \mu^{\text{pop}}$$

Effect of feed point location on population disequilibrium

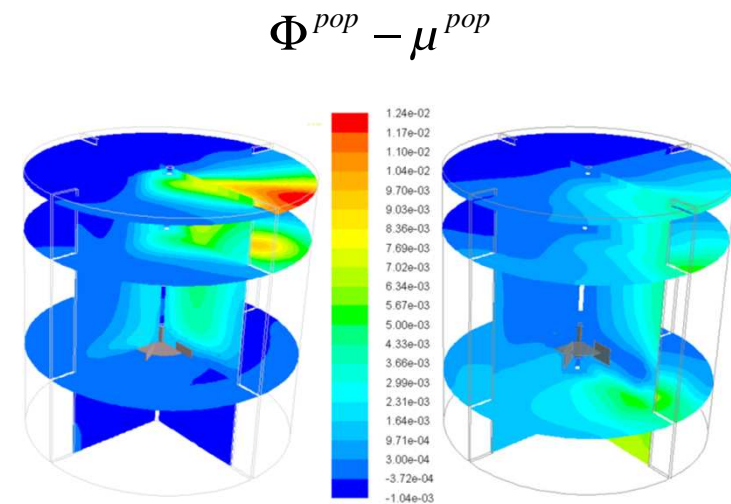
70 m³ aerated fermentor, kinetic reaction, G-L transfer

$\tau_M / \tau_S \rightarrow$ macroscopic gradients

μ^{pop} Spatially homogeneous

2°) Uptake ?

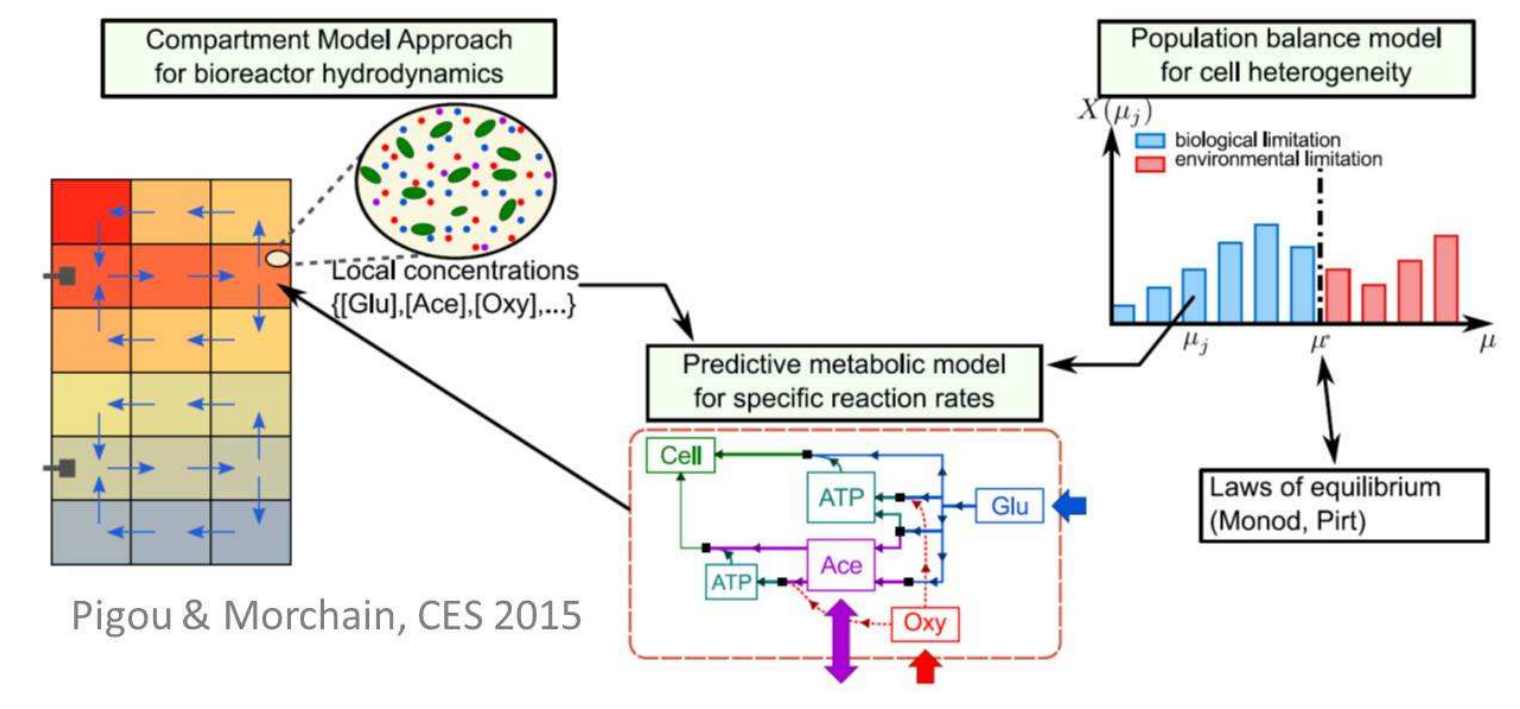
$Ta = T_{uptake} \sim T_{mix} \rightarrow \Phi^{pop}$ Spatially dependent



« Uptake rate-growth rate »

Local disequilibrium between Uptake rate-growth rate induce perturbation on cell metabolism : Consequences on bioreactions ???

Full Two-way coupling with « ADENON »



Investigating the interactions between physical and biological heterogeneities in bioreactors using compartment, population balance and metabolic models, Pigou and Morchain, CES, Accepted paper 2015

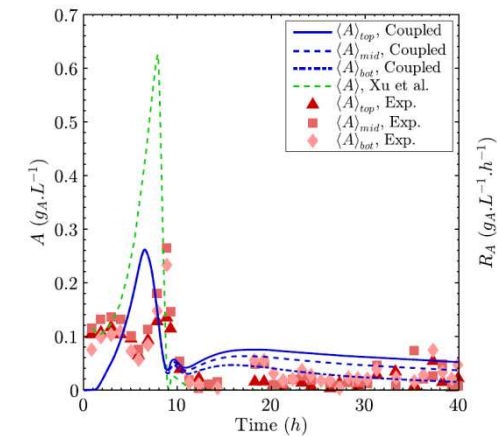
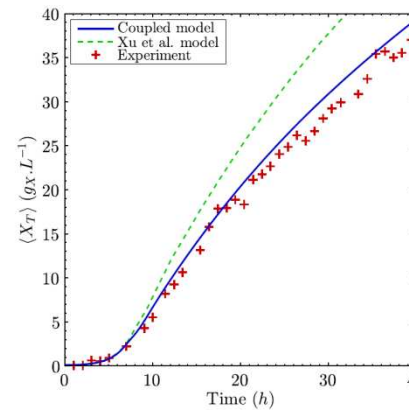
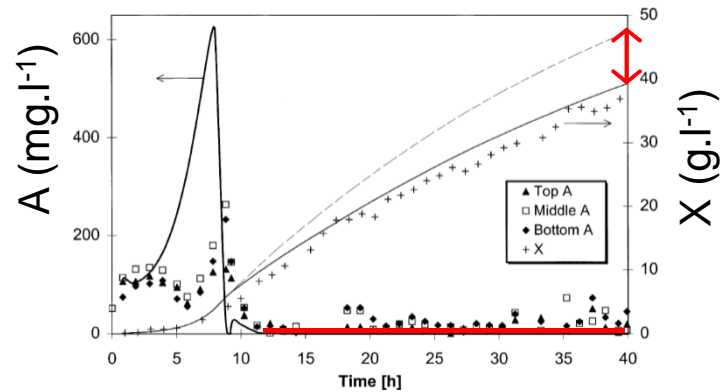


Full Two-way coupling with « ADENON »

ADENON : User Code developed by Morchain and Pigou

Batch and Fed-batch test cases from literature studies :

Compartment Model Approach (Vrâbel et al. 2001) + Metabolic model *E. coli* (Xu et al. 1999)



Accurate prediction of cell and by-product formation in time and space

Investigating the interactions between physical and biological heterogeneities in bioreactors using compartment, population balance and metabolic models, Pigou and Morchain, CES, Accepted paper 2015

Thank you for your attention

Questions and Answers

